This is a post-print of a paper published in the Bulletin of Engineering Geology and the Environment, in 2019. The paper was submitted on 20 June 2018, and this is the submitted version, as requested by the Publisher Policy.

Effect of UV radiation on chromatic parameters in serpentinites used as dimension stones.

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https://doi.org/10.1007/s10064-019-01469-3 Received: 20 June 2018 /Accepted: 9 January 2019 /Published online: 19 January 2019

# Bulletin of Engineering Geology and the Environment EFFECT OF UV RADIATION ON CHROMATIC PARAMETERS IN SERPENTINITES **USED AS DIMENSION STONE**

Manuscript Number:						
Full Title:	EFFECT OF UV RADIATION ON CHROMATIC PARAMETERS IN SERPENTINITES USED AS DIMENSION STONE					
Article Type:	Original Article					
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Funding Information:	Geosciences Centre of the University of Coimbra (UID/Multi/00073/2013)	Dra. Lidia Catarino				
	Fundação para a Ciência e a Tecnologia	Dra. Lidia Catarino				
	Spanish Science and Technology Ministry (BTE2003–04812)	Dra. Dolores Pereira Gomez				
	Spanish Science and Technology Ministry (CGL2004–03048)	Dra. Dolores Pereira Gomez				
	Spanish Science and Technology Ministry (CGL2005-03048/BTE)	Dra. Dolores Pereira Gomez				
	Spanish Science and Technology Ministry (CGL2006–05128/BTE)	Dra. Dolores Pereira Gomez				
	Spanish Science and Technology Ministry (CGL2010-18579/BTE)	Dra. Dolores Pereira Gomez				
	Spanish Geological Survey (IGME) (CONSTRUROCK)	Dr. Rafael Navarro Domínguez				
	General Foundation of the University of Salamanca (Doctor TCUE 2015)	Dr. Rafael Navarro Domínguez				
Abstract:	Colour is an important parameter that must be taken into account if aesthetic homogeneity in both construction and restoration work is to be achieved. In serpentinites, the different degrees of serpentinization or carbonation can influence their final colour. Sunlight, especially ultraviolet (UV) radiation, is a common agent that can decay properties such as hue, lightness, and chroma, and can have a very significant effect on the ageing process of dimension stones. Testing the effects of UV radiation is fundamental for predicting how materials may behave upon exposure. In this work, the surface alteration of materials due to UV radiation by means of arc-xence lamps was carried out to determine the degree of variation in colour parameters in serpentinites used as dimension stones. It was observed that the effect of UV radiation					

	on the materials studied was low, and that the samples tended to darken and become yellow. Gloss was the parameter that varied the most, which decreased in all cases analysed. Associations were observed between the loss of gloss and a decrease in some parameters such as open porosity, water absorption at atmospheric pressure and water absorption by capillarity (direct), bulk density and uniaxial compression strength (reverse). The mineralogical composition of the stone was the main factor that influenced the loss of gloss. This research highlights the importance of the detailed study of colour variation in dimension stones due to exposure to UV radiation. The incorrect assessment of the damage that could be caused by this agent can lead to severe aesthetic destruction that may result in expensive legal actions.
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# EFFECT OF UV RADIATION ON CHROMATIC PARAMETERS IN SERPENTINITES USED AS DIMENSION STONE

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#### Abstract

Colour is an important parameter that must be taken into account if aesthetic homogeneity in both construction and restoration work is to be achieved. In serpentinites, the different degrees of serpentinization or carbonation can influence their final colour. Sunlight, especially ultraviolet (UV) radiation, is a common agent that can decay properties such as hue, lightness, and chroma, and can have a very significant effect on the ageing process of dimension stones. Testing the effects of UV radiation is fundamental for predicting how materials may behave upon exposure. In this work, the surface alteration of materials due to UV radiation by means of arc-xenon lamps was carried out to determine the degree of variation in colour parameters in serpentinites used as dimension stones. It was observed that the effect of UV radiation on the materials studied was low, and that the samples tended to darken and become yellow. Gloss was the parameter that varied the most, which decreased in all cases analysed. Associations were observed between the loss of gloss and a decrease in some parameters such as open porosity, water absorption at atmospheric pressure and water absorption by capillarity (direct), bulk density and uniaxial compression strength (reverse). The mineralogical composition of the stone was the main factor that influenced the loss of gloss. This research highlights the importance of the detailed study of colour variation in dimension stones due to exposure to UV radiation. The incorrect assessment of the damage that could be caused by this agent can lead to severe aesthetic destruction that may result in expensive legal actions.

Keywords: Chroma, colour, dimension stone, gloss, lightness, serpentinite.

### 1. Introduction

The chromatic characteristics of building materials depend on several factors, including mineralogical composition, the presence of small amounts of iron or manganese oxides or sulphides, such as pyrite (Bams and Dewaele 2007), the textural aspects of rock (Esbert et al. 1997, Erdogan 2000) and properties such as porosity (Erdogan 2000) or finish (Benavente et al. 2003, Simonot and Elias 2003, Sanmartín et al. 2011). In dimension stones, colour homogeneity is one of the most important characteristics considered, depending on the application and use. This is especially important outdoors where these stones have been used to construct historic and more recent buildings, since one of its main functions is decorative. This is also a common problem in large structures, where the material used could come from more than one quarry (Sousa and Gonçalves 2012).

Polished finishes are more popular than other types of finishes, such as rustic, honed and bush hammered, because at this stage the surface becomes bright and the material reaches its maximum definition in terms of the textural aspects that enhance its chromatic qualities (Careddu and Marras 2013). Prolonged exposure to sunlight, the main source of ultraviolet radiation (UV), together with increases and decreases in temperature or humidity, can cause serious problems of decay. These include scaling (Sáez-Pérez and Rodríguez-Gordillo 2008), biodeterioration due to the presence of black fungi and cyanobacteria (Sterflinger 2011), soiling due to the combination with atmospheric particles from pollution (Urosevic et al. 2012) and variations in colour due to oxidation of some minerals, such as pyrite or chromite (Winkler 1997, Meierding 2005). Many white marbles are prone to discolouration, turning into more creamy or yellowish tones; grey limestones tend to increase in lightness and whitening due to solar action; and some brown-hued travertines tend to bleach due to prolonged exposure to sunlight (Winkler 1997; Harrell et al. 2007). Serpentinites, on the other hand, tend to lose stone fragments and form yellowish and reddish crusts due to the possible alteration of relict olivine, rich in iron, and the transformation of chromite to limonite (Meierding 2005). Although polishing accentuates the chromatic characteristics of the materials and gives a high degree of protection against damage caused by climate changes, it also highlights defects in the stone (variations of tones and textures, veins, etc.) (Ozcelik et al. 2012). The polished surface of a dimension stone tends to become opaque, and the gloss value decreases as a result of varying weather conditions, such as rain, wind, snow, temperature variation (Winkler 1979) or friction due to use (Blanchard 2012).

In the case of serpentinites, this fact is even more relevant, since their characteristics present large variations in colour, texture and mineralogy, from one outcrop to the next, due to differing degrees of serpentinization or carbonation (Pereira et al. 2007; 2013; Navarro et al. 2013; 2018). Therefore, colour is an important parameter that must be measured and controlled for achieving a certain degree of homogeneity in new construction and restoration works and in the replacement of damaged fragments; thus avoiding undesirable visual effects (Fig.1). Additionally, possible legal actions can be prevented in

cases where potential colour changes in already placed stones have not been previously discussed with the client. The standard specification for serpentine dimension stone, ASTM C1526-02 (2002), indicates that an alteration in the colour and finish of the serpentine may occur as a result of exposure to changing weather conditions, which should be taken into account in the evaluation of the suitability of serpentinite for a specific emplacement and weather context. The alteration of serpentinite, due to exposure to the weather, can result in whitish mineral deposits on the surface of the stone, discolouration, and loss of the polished surface.



Fig. 1. Unattractive visual appearance of serpentinite slabs with different chromatic characteristics placed on the exterior the Granada Exhibition and Convention Centre

UV radiation caused by sunlight is a common decay agent of the colour properties of materials (hue, lightness, and chroma), and can have a very significant effect on the ageing process of building materials (Careddu and Marras 2013; Sáez-Perez and Rodriguez-Gordillo 2008). Therefore, testing the effects of UV radiation is fundamental for predicting how materials may behave upon exposure. The surface alteration of serpentinite caused by exposure to UV radiation was carried out using arc-xenon lamps to determine the degree to which the colour parameters change after such exposure. The xenon-arc radiation source, in combination with a filter system, was designed to modify the distribution of the radiation.

Although colour measurement is a basic technique used in the field of conservation and restoration of dimension stone (Prieto et al. 2010), no method has been proposed for examining colour changes in the context of this study. Most methods involve paints and coatings, and there are different ISO and UNE standards that regulate their use. In addition, there are more studies on colour change produced after the application of protective coatings in restoration projects (e.g. Poli et al. 2006, Becherini et al. 2017, Carmona-Quiroga et al. 2018) or after the application of cleaning treatments (e.g. Grossi et al. 2007; Carmona-Quiroga et al. 2017; Pozo-Antonio et al. 2017; 2018). Moreover, there is a lack of publications that measure colour changes in stones. Only Careddu and Marras (2013) measure the effects of solar UV radiation on the gloss of marble and limestone samples. In this work, they use a similar method to the one used here, but instead filter radiation so that it mimics radiation coming through a glass window. Also, they apply 28 cycles of radiation in a 24-hour period.

The objective of this work was to determine variation in the chromatic characteristics of several types of polished serpentinites, extracted from the south of Spain, which occurs when UV radiation with specific wavelengths continuously strikes the surface of the stones. This study is particularly relevant because serpentinites from Granada (South-eastern Spain) are included as candidates as a Global Heritage Stone Resource (Cooper 2015, Marker 2015) and are found in many Spanish heritage buildings (Navarro et al. 2015). Likewise, the serpentinites from Macael (Almería, South-eastern Spain) are included, together with White Macael marble, as candidates as Global Heritage Stone Province, due to their frequent use in both recently built and historical buildings (Navarro 2016).

#### 2. Materials and Methodology

### 2.1 Materials

The research material comes from two different areas in Andalusia (Southern Spain) and both have been previously described in detailed (Navarro et al. 2018 and references therein). The quarries where the stones were sampled are located in Guejar Sierra (Granada) and in the Macael area (Almeria) (South-eastern Spain) (Fig. 2). The quarry in Granada, "Barranco de San Juan", ceased its activity during the middle of the 20<sup>th</sup> century. The four quarries in the Macael area are called "Virgen del Rosario", "C.E. Verde Almeria", "La Carrasca" and "La Milagrosa". The stone from the quarry in Granada is known as "Verde Granada" (VG) and was used in the construction of historic buildings such as the Alhambra, the Granada cathedral and the Royal Palace in Madrid (Navarro et al. 2015). In the Macael area, the stones are called "Verde Macael" (VM) from the quarry "Virgen del Rosario", "C.E. Verde Almeria", and samples CA and MI were taken from the quarries "La Carrasca" and "La Milagrosa", respectively. Both of these quarries have since been abandoned and there is no information about the commercial names of the stones extracted from them.

Geologically, these serpentinites are placed in the Nevado-Filábride Complex (Internal Zone of the Betic Cordillera) (Fig. 2), which can be subdivided into lower and upper units (Martín-Algarra et al. 2004). The serpentinites quarries are located in the upper unit. From the bottom to the top, this upper unit is composed of metaconglomerates, light micaschists and quartzites, dark schists and marble levels, with metamorphosed lenses of basic and ultrabasic rocks, totally or partially transformed to eclogites, amphibolites, and serpentinites interspersed throughout the sequence. The age of this unit is from Palaeozoic to Triassic (Gómez-Pugnaire et al. 2000; Puga et al. 2002).



**Fig. 2.** Location of the quarries sampled: 1. Barranco de San Juan (VG); 2. Virgen del Rosario (VM); 3. C.E. Verde Almería (VA); 4. La Carrasca (CA). 5. La Milagrosa (MI) (Modified from Vera 2004)

Two polished tiles with the dimensions of 120 x 60 x 25 mm were studied from each serpentinite sample extracted. Two samples were selected from the stone type Verde Granada (VG) due to their differences in hue, light green (VG1) and dark green (VG2). Two samples with different carbonate contents (VM1 and VM2) of the stone type "Verde Macael" were studied, as well as the stone type Verde Almería (VA) and the two samples taken from La Carrasca (CA) and La Milagrosa (MI). The samples VG1, VG2 and VA were prepared at the Los Carriles Marble Handicraft Workshop in Macael, VM1 and VM2 were polished by the company "Mármoles Naturales, SL" and samples CA and MI were polished at the laboratory of the Spanish Geological Survey in Madrid.

Fig. 3 (a-g) shows the appearance of the different polished samples before testing. Fig. 3h shows schematically the location of the measured points.



Fig. 3. Appearance of the polished samples before the test: a) Verde Granada 1 (VG1); b) Verde Granada 2 (VG2); c) Verde Macael 1 (VM1); d) Verde Macael 2 (VM2); e) Verde Almería (VA); f) La Carrasca (CA); g) La Milagrosa (MI); h) Location of the points measured. Scale bar = 1 cm.

The mineralogy of the samples studied is shown in Table 1 and the physical and mechanical properties of these serpentinites are shown in Table 2 (Navarro et al. 2013, 2018):

Sample	Main	Accessory	Secondary
VG1 <sup>+</sup>	VG1 <sup>+</sup> Atg Ol, Px, Am, Chl, Cb, Crt, Mag, Py, Sp, Ilm, Hem		Cb, Tlc, Trm
VG2 <sup>+</sup>	Atg	Ol, Px, Am, Chl, Cb, Crt, Mag, Py, Sp, Ilm, Hem	Cb, Tlc, Trm
$VM1^+$	Cb	Atg, Px, Amp, Mag, Py, Ttn, Sp	Cb, Hmt
$VM2^+$	Atg, Cb	Ol, Px, Chl, Crt, Mag, Ilm, Py, Sp	Cb, Ilm, Hem
$VA^+$	Atg	Cb, Crt, Chl, Mag, Py, Ilm, Hem, Grt	Cb, Tlc, Hem
$CA^+$	Px; Amp; Ep	Atg, Crt, Chl, Mag, Ilm, Chr, Hem	Cb, Tlc, Hem
MI <sup>x</sup>	Atg	Am, Crt, Chl, Mag, Ilm	Cb, Tlc, Hem

Table 1: Main mineral contents of the serpentinites studied (<sup>+</sup>Navarro et al. 2018; <sup>x</sup> Navarro et al. 2013). Abbreviations: Atg: antigorite; Am: amphibole; Cb: carbonates; Chl: chlorite; Chr: chromite; Crt: chrysotile; Ep: epidote; Grt: garnet; Hem: hematites; Ilm: ilmenite; Mag: magnetite; Ol: olivine; Px: pyroxene; Py: pyrite; Sp: sphalerite; Tlc: talc; Trm: tremolite; Ttn: titanite; (Abbreviations: Siivola and Schmid 2007)

Sample	BD (kg/m <sup>3</sup> )	OP (%)	WAA (%)	WCC (g/m <sup>2</sup> s <sup>0.5</sup> )	V <sub>p</sub> (km/s)	CS (MPa)	FS (MPa)	KM (MPa)
VG1 <sup>+</sup>	$2666\pm27$	$0.53\pm0.36$	$0.19\pm0.12$	$0.66\pm0.59$	$5.65\pm0.48$	331 ± 41	34 ± 7	2398 ± 1123
VG2 <sup>+</sup>	$2658\pm9$	$0.92\pm0.28$	$0.33\pm0.07$	$1.17\pm0.49$	$5.53\pm0.44$	$361 \pm 60$	$19\pm9$	$3198 \pm 1115$
VM1+	$2888 \pm 59$	$0.27\pm0.11$	$0.10\pm0.03$	$0.26\pm0.13$	$6.13\pm0.43$	$139\pm42$	23 ± 10	$1928\pm721$
VM2+	2704 ± 5	$0.34\pm0.07$	$0.16\pm0.02$	$0.40\pm0.09$	$6.02\pm0.19$	$315\pm43$	14 ± 15	$3235 \pm 1432$
$VA^+$	$2650\pm11$	$1.37\pm0.39$	$0.53\pm0.11$	$2.19 \pm 1.05$	$5.38\pm0.45$	$246\pm37$	27 ± 12	$1150\pm319$
$CA^+$	$2949 \pm 256$	$1.31\pm0.14$	$0.42\pm0.08$	*	$5.03\pm0.89$	$279\pm84$	22 ± 3	$790 \pm 106$
MI <sup>x</sup>	2687 ± 17	$0.73 \pm 0.24$	$0.30 \pm 0.08$	*	$5.76 \pm 0.51$	252 ± 34	*	$1631 \pm 583$

Table 1: Main physical and mechanical parameters (mean and standard deviation) (<sup>+</sup>Navarro et al. 2018; <sup>x</sup> Navarro et al. 2013). Abbreviation: BS: Bulk density; OP: Open porosity; WAP: Water absorption at atmospheric pressure; WCC: Water absorption coefficient by capillarity; Vp: velocity of the ultrasonic p-waves; CS: Compressive strength; FS: Flexural strength; KM: Knoop microhardness. \* Not tested.

## 2.2 Methodology

A Q-Lab climatic chamber at the Faculty of Science and Technology of the University of Coimbra (Portugal) (Q Sun Xenon Test Model Xe-3-HS) was used to simulate the possible effect that continuous exposure to UV Rays, with specific wavelengths, had on the chromatic parameters of polished serpentinite samples prepared for commercial use. The test conditions were carried out following the ISO 11341: 2004, which is standard that best fits. The temperature, humidity and radiation values were selected taking into account the climatic characteristics of the city of Granada (Spain). The annual average global solar radiation in the city, according to Sancho-Ávila et al. (2012) is 5.20 kW\*h/m<sup>2</sup>\*day. The average relative humidity varies between 40% and 80% and the maximum average temperature is 34°C (INM 2004). The weather station "Granada Base Aerea" (Latitude: 37° 8'13" N - Longitude: 3° 37' 53" W - Height: 687 masl) was the reference.

The samples were irradiated by the UV lamps during 4 hours, with an irradiance of  $0.63 \text{ W} / \text{m}^2 * \text{per}$  every wavelength (in nm), which is the result of the conversion of the average global radiation value in Granada. The ambient temperature was fixed at 40°C, according to the ISO 11341 (2004) Standard and the relative humidity was fixed at 40%. After this treatment, the samples passed through a condensation step for 4 hours, at which time no UV radiation was applied. The ambient temperature was 40 °C and the relative humidity was 80%. The combination of these two steps constituted a cycle. The samples were exposed to 126 cycles and the total duration of the experiment was 1008 hours.

A Konica Minolta CM700d spectrophotometer from the Faculty of Science and Technology of the University of Coimbra (Portugal) was used to measure colour parameters in the specular component excluded mode (SCE). This decision was based on the work by Sanmartin (2012), who analysed the influence of the inclusion or exclusion of the specular component in the measurement mode on granites with different finishes (heterogeneous rocks from the point of view of colour). She found that the excluded specular component (SCE mode) magnifies the differences in colour produced by the different surface finishes, making this the preferred mode when analysing colour differences, as opposed to the specular component included mode (SCI). The illuminant mode used was 8°/D65, with a measuring window 8 mm in diameter (Medium Area View-MAV). The value 8° refers to the angle with respect to the position of the observer, while the illuminant D65 is a standard illuminant (daylight, colour temperature: 6504 K). The colour space used was based on the tristimulus values, L\*a\*b\*, of the 1976 CIEL\*a\*b\* colour system (Smith 1997) (Fig. 4). The data obtained were expressed by an alphanumeric code that represents the colour coordinates, that is to say the three axes of the three-dimensional CIEL\*a\*b\* system, of the exact point that is being measured (Fig. 4). The vertical axis is L\* (lightness) with a range of values from black (0) to white (100). The horizontal plane is defined by axes a\* and b\*. The a-axis varies between red (+60) and green (-60) and the b-axis varies between yellow (+60) and blue (-60).

The colour of the sample (chroma) is defined by the chromatic coordinates, C \*, whose value is determined according to the expression (CIE 2004) (1):

$$C^* = \sqrt{(a^*)^2 + (b^*)^2} \tag{1}$$

The hue angle  $(h_{ab})$  is defined as (2):

$$h_{a^*b^*} = \arctan(b^*/a^*) \tag{2}$$

The total colour difference is the Euclidean distance between two points within it. It is denoted as  $\Delta E$  and its equation is (CIE 2004) (3):

$$\Delta E = \sqrt{(\Delta L^*)^2 + (\Delta a^*)^2 + (\Delta b^*)^2}$$
(3)

The differences in hue  $\Delta H^*$  is calculated according to the following formula (4):

$$\Delta H^* = \sqrt{(\Delta a^*)^2 + (\Delta b^*)^2 - (\Delta C^*)^2} \tag{4}$$

Hue is positive if the hue angle h of the sample is greater than the reference angle and negative if the hue angle of the sample is smaller.



Fig. 4. Representation of the 1976 CIEL\*a\*b\* space (modified from Smith 1997).

In addition, this spectrophotometer incorporates a reflectometer that allows the gloss value, expressed in GU (gloss units) at an angle of 8°, to be obtained. The gloss is defined as the ratio between the amount of light reflected from a surface and the amount of light incident on that surface (Cayless et al. 2011). This spectrophotometer measures the amount of light that is reflected when a ray of light hits the surface at a certain angle, which in this case was 8°. The values range from 0 (opacity) to 100 (maximum gloss) GU.

The mean values were obtained from a total of thirteen points, with the three measurements taken at each point (Fig. 3 h). The position of the spectrophotometer was fixed by a mask to measure always at the same points, and the values were measured at the start and at the end of the experiment.

# 3. Results

After carrying out 126 irradiation cycles, the colour parameters of the samples tested were measured and compared to the initial values recorded. Regarding the initial colour measurements, two groups with similar characteristics were established. It was observed that most of the samples presented values of L\* that were around 23, ranging between 19.36 and 27.82. However, there were two samples, VG1 and CA, with a lighter colour, with L\* values equal to 39.57 and 47.55, respectively, and somewhat higher values for parameters a\* and b\*. Table 3 lists the initial values and Table 4 the final values.

Sample	L*	a*	b*	C*	h <sub>ab</sub> (°)	G <sup>8</sup> (GU)
VG1	$39.57 \pm 4.61$	$-6.70 \pm 1.55$	$2.67 \pm 1.58$	$7.21 \pm 1.82$	$-21.75 \pm 13.67$	$75.08 \pm 7.69$
VG2	$23.33 \pm 2.53$	$-3.09 \pm 1.58$	$1.73 \pm 1.60$	$3.54 \pm 1.97$	$-29.33 \pm 24.27$	$63.96 \pm 6.43$
VM1	$27.82 \pm 2.59$	$-5.66\pm0.78$	$0.59\pm0.50$	$5.69\pm0.81$	$-5.94 \pm 4.91$	$65.69 \pm 5.72$
VM2	$22.19\pm5.16$	$-3.69\pm0.98$	$-0.21\pm0.45$	$3.70\pm0.97$	$3.27\pm6.70$	$63.04 \pm 8.47$
VA	$19.36\pm3.45$	$-2.92 \pm 1.01$	$\textbf{-0.86} \pm 0.55$	$3.04\pm0.88$	$16.37 \pm 13.98$	$62.65 \pm 10.64$
CA	47.55 ± 2.88	$-6.11 \pm 0.62$	$1.67 \pm 1.18$	$6.33\pm0.82$	$-15.26 \pm 9.48$	$71.58 \pm 6.84$
MI	$24.15\pm3.37$	$-2.46\pm0.52$	$\textbf{-1.19} \pm 0.78$	$2.73\pm0.49$	$25.79 \pm 16.56$	$80.08 \pm 7.96$

Table 3. Initial mean values and standard deviation of the main colour parameters.

Sample	L*	a*	b*	C*	h <sub>ab</sub> (°)	G <sup>8</sup> (GU)
VG1	$38.56 \pm 4.46$	$-6.53 \pm 1.38$	$2.57 \pm 1.52$	$7.02 \pm 1.68$	$-21.47 \pm 12.13$	$70.85\pm9.37$
VG2	$23.37\pm2.45$	$-2.97 \pm 1.57$	$1.79 \pm 1.55$	$3.47\pm2.05$	$-31.12 \pm 16.81$	$60.15\pm5.53$
VM1	$26.98 \pm 2.45$	$-5.69\pm0.71$	$0.81\pm0.42$	$5.75\pm0.74$	$-8.12 \pm 3.63$	$63.35\pm4.82$
VM2	$22.14\pm5.30$	$-3.92\pm0.78$	$0.06\pm0.35$	$3.92\pm0.80$	$-0.83 \pm 4.59$	$62.58 \pm 5.40$
VA	$19.48\pm3.16$	$-2.67\pm0.93$	$\textbf{-0.61} \pm 0.50$	$2.74\pm0.83$	$12.97\pm13.82$	$56.12\pm9.86$
CA	$46.55\pm3.06$	$\textbf{-6.01} \pm 0.62$	$1.69 \pm 1.16$	$6.24\pm0.82$	$-15.71 \pm 9.58$	$70.96\pm5.83$
MI	23.83 ± 3.25	$-1.88 \pm 0.54$	$-1.04 \pm 0.87$	$2.15\pm0.64$	$28.85 \pm 22.42$	$74.58 \pm 7.46$

Table 4. Final mean values and standard deviation of the main colour parameters after 126 cycles.

From these values, the variation of lightness ( $\Delta L^*$ ), chromatic coordinates ( $\Delta a^*$  and  $\Delta b^*$ ), Chroma ( $\Delta C^*$ ), gloss ( $\Delta G8$ ) were calculated. The colour difference ( $\Delta E$ ) and the hue difference ( $\Delta H$ ) were also obtained (Table 5).

Sample	$\Delta L^*$	∆a*	$\Delta b^*$	$\Delta C^*$	Δh	$\Delta G^{8} (GU)$	$\Delta E_{ab}$	ΔΗ
VG1	-1.02	0.17	-0.10	-0.19	0.28	-4.23	1.04	0.03
VG2	0.04	0.12	0.06	-0.07	-1.79	-3.81	0.14	0.11
VM1	-0.84	-0.03	0.22	0.05	-2.18	-2.35	0.87	-0.22
VM2	-0.04	-0.22	0.27	0.22	-4.10	-0.46	0.35	-0.27
VA	0.11	0.25	0.24	-0.30	-3.40	-6.54	0.37	-0.17
CA	-1.00	0.10	0.02	-0.09	-0.44	-0.62	1.01	-0.05
MI	-0.32	0.58	0.15	-0.59	3.06	-5.50	0.68	0.13

Table 5. Variation of the mean values after 126 cycles.

# 4. Discussion

The effect of the ageing process due to UV radiation on the materials studied was low. In the CIEL\*a\*b\* colour space, variations in lightness and Chroma lower than 3 units are imperceptible to the human eye (Völz 2001). In this study, none of the samples exhibited variations greater than this value. However, all the samples, except for VA and VG2, tended to darken, meaning that the values obtained for lightness (L\*) were lower. The values obtained for lightness (L\*) in samples VA and VG2, on the other hand, varied only slightly. The chromatic coordinates a\* and b\* shifted towards yellow, and the Chroma (C\*) and the hue (h) tended to decrease. The variation observed with respect to the other parameters, such as the total colour difference ( $\Delta E_{ab}$ ) and the difference in hue ( $\Delta H$ ), was also minimal. In some cases, especially with regard to the h<sub>ab</sub> parameter, standard deviation was quite high, which may be the result of small differences in hue at the positions measured, owing to the presence of minerals with different hue characteristics. The variation of these parameters could be associated with the presence of some minerals, such as pyrite and iron oxides that, even in small quantities, can produce staining upon oxidation in wet conditions, modifying the colour characteristics (Bams and Dewaele, 2007). Other authors (Winkler 1997) have reported that sunlight can modify lightness due to bleaching and oxidation processes.

Gloss was the parameter that exhibited the most variation, as it had decrease in all of the samples. In the case of the VG1 sample, variation was -4.23 GU (5.64%), while in the darkest sample (VG2) it was -3.81

GU (5.95%). In the most carbonated samples, VM1 and VM2, the variation in gloss was of -2.35 GU (3.57%) and -0.46 GU (0.73%) respectively. In the VA variety, which had the largest decrease in lightness, the decrease in gloss was found to be -6.54 GU (10.44%). The samples of the variety CA varied only -0.62 GU (0.86%). Finally, in the MI samples, this variation was -5.50 GU (6.87%). Also, it was observed that the samples with greater mineralogical diversity had less gloss loss, which was less than 4%. This was the case of VM and CA, which contain other components such as carbonates, pyroxenes and amphiboles or antigorite. However, the samples which were comprised almost exclusively of antigorite (VG1 and VG2, VA and MI) suffered gloss losses ranging from 4% to 11%. It should be noted that the loss of gloss is caused by the erosion of a small amount of material on the surface of the stone, and that the loss of several hundred nanometers of surface material is sufficient enough to affect gloss. Furthermore, it is likely that the mechanism of erosion involves superficial chemical and mechanical processes (Wood et al. 2000), and different mineralogical compositions of the stone can lead to different results.

Some authors, such as Erdogan (2000), have related the value of gloss with physical parameters like porosity, the presence of fractures and intergranular contacts. Also, other authors such as Ozcelik et al. (2012) or Careddu and Marras (2013) relate variation in gloss in decay tests with certain textural characteristics, such as grain size, as well as other physical properties such as density, absorption and even compression strength. Only Careddu and Marras (2013) have assessed changes in gloss due to the action of UV rays on marbles and limestones used as dimension stones. In the present work, the relationship between different physico-mechanical parameters and variations in gloss has been established. The results obtained are shown in Fig. 5.

With respect to physical properties, there is a direct relationship between the loss of gloss and parameters such as open porosity, water absorption at atmospheric pressure and water absorption coefficient by capillarity. As the values for these parameters increases, so did gloss loss. However, the relationship between gloss and bulk density and ultrasonic velocity (VP) was the opposite. The loss of gloss decreased as the samples became denser. This was observed for all samples except for the CA variety, which showed different results with respect to water absorption at atmospheric pressure and water absorption coefficient by capillarity. In addition, the VM2 sample had different results with respect to bulk density. It is remarkable that the sample CA was the one that exhibited the most variation. This may be due to the fact that it has greater mineralogical diversity, and is a rock with incomplete serpentinization (Navarro et al., 2018), causing this stone to present behaviour that is different from serpentinites *sensu strict*.

With respect to the relationship between the loss of gloss and mechanical properties, an inverse relation was observed when gloss was compared to the compression strength of the sample. The loss of gloss decreased as the value of compressive strength increased. Based on this relationship, we were able to divide the samples into two groups. The first group included the samples with the highest carbonate content and the greatest mineralogical diversity such as VM1, VM2, and CA. These were the samples with the least amount of gloss loss (less than 4%). The second group included the samples comprised





Fig. 1. Correlations between the different physical and mechanical parameters and loss of gloss.

No relationship was observed between the loss of gloss and flexural strength. Also, the differences that were observed for compressive strength, due to mineralogical composition, were not observed for flexural strength. With respect to the Knoop microhardness, a certain inverse relationship was also observed. The

samples exhibiting higher values for this parameter had less loss of gloss, which was the case for all of the samples analysed, except for the CA variety. In this sample the results were different, being similar to what was observed for the physical properties. However, it should be noted that the correlation coefficient was relatively low in this case.

Density, water absorption at atmospheric pressure, compression strength and flexural strength did not influence the gloss. Careddu and Marras (2013) indicate that only porosity and grain size are related to the loss of gloss in limestones and marbles. Our findings are in agreement with this, as it was observed that porosity, water absorption by capillarity and compression strength were the parameters with the best correlation. This is due to the fact that during the polishing treatment, the pores are sealed in the most superficial layer, increasing the gloss (Primavori 2006). This superficial layer will later be the one that will be most altered due to the action of ultraviolet rays, due to the change of its physical properties (Careddu and Marras 2013).

### 5. Conclusions

After carrying out the accelerated ageing experiment by applying UV radiation with an arc-xenon lamp onto polished samples of serpentinite it is observed that:

- The main parameters that define colour in the CIE L\*a\*b\* space showed very little variation. In general, it was possible to observe a decrease in lightness, a tendency towards turning yellow and a decrease in colour saturation (Chroma).
- The greatest variation was observed in the gloss, which decreased within the range of 0.46 GU (0.73%) to -6.54 GU (10.44%) and was lower in the samples with a more diverse mineralogical composition.
- This loss of gloss was generally well correlated with physical parameters such as bulk density, porosity, absorption at atmospheric pressure and absorption by capillarity.

This work highlights the importance of the correct characterization of this type of material. The determination of the variation of colour caused by UV radiation is an aspect that is not usually taken into consideration when placing materials outdoors. The orientation in which the materials are placed determines how long they are exposed to sunlight and UV radiation. Thus, the colour of the stone can vary from one area to another, as well as aesthetic damage. In some cases the application of conservation treatments and even the replacement of certain pieces are sometimes necessary, which can result in economic losses. Additionally, determining colour in dimension stones and its variation due to UV radiation is essential in the study of the application of chemical and biological consolidating treatments, coatings or the application of cleaning processes that may damage the stone.

However, this topic has not been studied in detail and additional research is necessary for establishing how these treatments may interact with the properties of the stone. The possible alteration of colour due to different climatic conditions (intensity of UV radiation, time of exhibition or humidity), different

finishes (e.g. polished, hammered or brushed) and materials is an area that could lead to new research opportunities.

#### Acknowledgements

This work was made possible thanks to the FCT (Fundação para a Ciência e a Tecnologia, I.P.) and by the research project UID/Multi/00073/2013 of the Geosciences Centre of the University of Coimbra (Portugal), Spanish Science and Technology Ministry (MCT) through the projects: BTE2003–04812, CGL2004–03048, CGL2005-03048/BTE, CGL2006–05128/BTE and CGL2010-18579/BTE, Spanish Geological Survey (I.G.M.E.) through the project CONSTRUROCK (Natural stone and their relation with historic and monumental heritage and new building network) and the General Foundation of the University of Salamanca through the call Doctor TCUE 2015.

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Sample	Main	Accessory	Secondary
VG1 <sup>+</sup>	Atg	Ol, Px, Am, Chl, Cb, Crt, Mag, Py, Sp, Ilm, Hem	Cb, Tlc, Trm
VG2 <sup>+</sup>	Atg	Ol, Px, Am, Chl, Cb, Crt, Mag, Py, Sp, Ilm, Hem	Cb, Tlc, Trm
VM1 <sup>+</sup>	Cb	Atg, Px, Amp, Mag, Py, Ttn, Sp	Cb, Hmt
VM2+	Atg, Cb	Ol, Px, Chl, Crt, Mag, Ilm, Py, Sp	Cb, Ilm, Hem
VA <sup>+</sup>	Atg	Cb, Crt, Chl, Mag, Py, Ilm, Hem, Grt	Cb, Tlc, Hem
CA <sup>+</sup>	Px; Amp; Ep	Atg, Crt, Chl, Mag, Ilm, Chr, Hem	Cb, Tlc, Hem
MI <sup>x</sup>	Atg	Am, Crt, Chl, Mag, Ilm	Cb, Tlc, Hem

Table 1: Main mineral contents of the serpentinites studied (<sup>+</sup>Navarro et al. 2018; <sup>x</sup> Navarro et al. 2013). Abbreviations: Atg: antigorite; Am: amphibole; Cb: carbonates; Chl: chlorite; Chr: chromite; Crt: chrysotile; Ep: epidote; Grt: garnet; Hem: hematites; Ilm: ilmenite; Mag: magnetite; Ol: olivine; Px: pyroxene; Py: pyrite; Sp: sphalerite; Tlc: talc; Trm: tremolite; Ttn: titanite; (Abbreviations: Siivola and Schmid 2007)

Sample	BD (kg/m <sup>3</sup> )	OP (%)	WAA (%)	WCC (g/m <sup>2</sup> s <sup>0.5</sup> )	V <sub>p</sub> (km/s)	CS (MPa)	FS (MPa)	KM (MPa)
$VG1^+$	$2666 \pm 27$	$0.53\pm0.36$	$0.19\pm0.12$	$0.66\pm0.59$	$5.65\pm0.48$	331 ± 41	$34 \pm 7$	$2398 \pm 1123$
$VG2^+$	$2658\pm9$	$0.92\pm0.28$	$0.33\pm0.07$	$1.17\pm0.49$	$5.53\pm0.44$	$361\pm60$	$19\pm9$	$3198 \pm 1115$
$VM1^+$	$2888 \pm 59$	$0.27\pm0.11$	$0.10\pm0.03$	$0.26\pm0.13$	$6.13\pm0.43$	$139\pm42$	23 ± 10	$1928\pm721$
$VM2^+$	$2704\pm5$	$0.34\pm0.07$	$0.16\pm0.02$	$0.40\pm0.09$	$6.02\pm0.19$	$315\pm43$	14 ± 15	$3235 \pm 1432$
$VA^+$	$2650\pm11$	$1.37\pm0.39$	$0.53\pm0.11$	$2.19 \pm 1.05$	$5.38\pm0.45$	$246\pm37$	27 ± 12	$1150\pm319$
$CA^+$	$2949 \pm 256$	$1.31 \pm 0.14$	$0.42\pm0.08$	*	$5.03\pm0.89$	$279\pm84$	22 ± 3	$790 \pm 106$
MI <sup>x</sup>	$2687 \pm 17$	$0.73\pm0.24$	$0.30 \pm 0.08$	*	$5.76\pm0.51$	252 ± 34	*	1631 ± 583

Table 1: Main physical and mechanical parameters (mean and standard deviation) (<sup>+</sup>Navarro et al. 2018; <sup>x</sup> Navarro et al. 2013). Abbreviation: BS: Bulk density; OP: Open porosity; WAP: Water absorption at atmospheric pressure; WCC: Water absorption coefficient by capillarity; Vp: velocity of the ultrasonic p-waves; CS: Compressive strength; FS: Flexural strength; KM: Knoop microhardness. \* Not tested.

Sample	L*	a*	b*	C*	h <sub>ab</sub> (°)	G <sup>8</sup> (GU)
VG1	$39.57 \pm 4.61$	$-6.70 \pm 1.55$	$2.67 \pm 1.58$	$7.21 \pm 1.82$	$-21.75 \pm 13.67$	$75.08 \pm 7.69$
VG2	$23.33 \pm 2.53$	$-3.09 \pm 1.58$	$1.73 \pm 1.60$	$3.54 \pm 1.97$	$-29.33 \pm 24.27$	$63.96 \pm 6.43$
VM1	$27.82 \pm 2.59$	$-5.66\pm0.78$	$0.59\pm0.50$	$5.69\pm0.81$	$-5.94 \pm 4.91$	$65.69 \pm 5.72$
VM2	$22.19\pm5.16$	$-3.69\pm0.98$	$\textbf{-0.21} \pm 0.45$	$3.70\pm0.97$	$3.27\pm6.70$	$63.04 \pm 8.47$
VA	$19.36\pm3.45$	$-2.92 \pm 1.01$	$\textbf{-0.86} \pm 0.55$	$3.04\pm0.88$	$16.37\pm13.98$	$62.65 \pm 10.64$
CA	$47.55\pm2.88$	$\textbf{-6.11} \pm 0.62$	$1.67 \pm 1.18$	$6.33\pm0.82$	$-15.26 \pm 9.48$	$71.58 \pm 6.84$
MI	24.15 ± 3.37	$-2.46 \pm 0.52$	$-1.19 \pm 0.78$	$2.73\pm0.49$	$25.79 \pm 16.56$	80.08 ± 7.96

Table 3. Initial mean values and standard deviation of the main colour parameters.

Sample	L*	a*	b*	C*	h <sub>ab</sub> (°)	G <sup>8</sup> (GU)
VG1	$38.56 \pm 4.46$	$-6.53 \pm 1.38$	$2.57 \pm 1.52$	$7.02 \pm 1.68$	$-21.47 \pm 12.13$	$70.85\pm9.37$
VG2	$23.37 \pm 2.45$	$-2.97 \pm 1.57$	$1.79 \pm 1.55$	$3.47\pm2.05$	$-31.12 \pm 16.81$	$60.15\pm5.53$
VM1	$26.98 \pm 2.45$	$-5.69\pm0.71$	$0.81 \pm 0.42$	$5.75\pm0.74$	$-8.12 \pm 3.63$	$63.35\pm4.82$
VM2	$22.14 \pm 5.30$	$-3.92\pm0.78$	$0.06\pm0.35$	$3.92\pm0.80$	$-0.83 \pm 4.59$	$62.58 \pm 5.40$
VA	$19.48\pm3.16$	$-2.67\pm0.93$	$-0.61\pm0.50$	$2.74\pm0.83$	$12.97 \pm 13.82$	$56.12\pm9.86$
CA	$46.55\pm3.06$	$-6.01\pm0.62$	$1.69 \pm 1.16$	$6.24\pm0.82$	$-15.71 \pm 9.58$	$70.96 \pm 5.83$
MI	$23.83 \pm 3.25$	$-1.88 \pm 0.54$	$-1.04\pm0.87$	$2.15\pm0.64$	$28.85 \pm 22.42$	$74.58 \pm 7.46$

Table 4. Final mean values and standard deviation of the main colour parameters after 126 cycles.

Sample	$\Delta L^*$	∆a*	$\Delta b^*$	$\Delta C^*$	Δh	$\Delta G^{8} (GU)$	$\Delta E_{ab}$	ΔΗ
VG1	-1.02	0.17	-0.10	-0.19	0.28	-4.23	1.04	0.03
VG2	0.04	0.12	0.06	-0.07	-1.79	-3.81	0.14	0.11
VM1	-0.84	-0.03	0.22	0.05	-2.18	-2.35	0.87	-0.22
VM2	-0.04	-0.22	0.27	0.22	-4.10	-0.46	0.35	-0.27
VA	0.11	0.25	0.24	-0.30	-3.40	-6.54	0.37	-0.17
CA	-1.00	0.10	0.02	-0.09	-0.44	-0.62	1.01	-0.05
MI	-0.32	0.58	0.15	-0.59	3.06	-5.50	0.68	0.13

Table 5. Variation of the mean values after 126 cycles.